Synthesis and DNA Cleavage Activity of a Novel Bleomycin A₅ Glycoconjugate

Ambar K. Choudhury, Zhi-Fu Tao, and Sidney M. Hecht*,†

Departments of Chemistry and Biology, University of Virginia, Charlottesville, Virginia 22901

sidhecht@virginia.edu

Received January 23, 2001

ABSTRACT



To explore the possibility of modifying bleomycin in a fashion that could alter its physiological distribution in a therapeutic setting, a new analogue of bleomycin has been prepared. This analogue is intended to target the asialoglycoprotein receptor on liver cells. Critically, despite the large C-substituent, the bleomycin conjugate was found to degrade DNA in the same fashion as bleomycin A_5 itself, and with only modestly decreased efficiency.

The bleomycins (BLMs) are glycopeptide-derived antitumor agents¹ used clinically for the treatment of several neoplasms;² the antitumor activity of these agents is believed to result from their ability to mediate the selective cleavage of DNA³ and possibly also RNA.⁴ While the mechanisms of polynucleotide recognition and cleavage have been studied

intensively,^{3,4} much less is known about the basis for the selectivity of BLM action as a therapeutic agent. Factors that may influence therapeutic selectivity include selectivity of organ distribution⁵ and selective catabolism of the agent by bleomycin hydrolase.^{2,6} It is interesting that the cytotoxicity of bleomycin has been reported to be limited by low efficiency of cellular uptake; electropermeabilization dramatically increased the cytotoxicity of the drug.⁷

The asialoglycoprotein receptor,⁸ a specific membranebound receptor for glycoproteins in mammalian liver cells, recognizes the terminal galactose residues of synthesized

[†] Address correspondence to this author at the Department of Chemistry, University of Virginia, Charlottesville, VA 22901.

^{(1) (}a) Hecht, S. M. In *Cancer Chemotherapeutic Agents*; Foye, W. O., Ed.; American Chemical Society: Washington, DC, 1995; pp 369–388. (b) Hecht, S. M. *J. Nat. Prod.* **2000**, *63*, 158.

⁽²⁾ *Bleomycin Chemotherapy*; Sikic, B. I., Rozencweig, M., Carter, S. K., Eds.; Academic Press: Orlando, FL, 1985.

^{(3) (}a) Hecht, S. M. Acc. Chem. Res. **1986**, *19*, 83. (b) Kozarich, J. W.; Stubbe, J. Chem. Rev. **1987**, 87, 1107. (c) Natrajan, A.; Hecht, S. M. In Molecular Aspects of Anticancer Drug–DNA Interactions; Neidle, S., Waring, M. J., Eds.; Macmillan: London, 1980; pp 197–242. (d) Kane, S. A.; Hecht, S. M. Prog. Nucl. Acid Res. Mol. Biol. **1994**, *49*, 313. (e) Burger, R. M. Chem. Rev. **1998**, *98*, 1153.

^{(4) (}a) Hecht, S. M. *Bioconjugate Chem.* **1994**, *5*, 513. (b) Holmes, C. D.; Duff, R. J.; van der Marel, G. A.; van Boom, J.; Hecht, S. M. *Bioorg. Med. Chem.* **1997**, *5*, 1235. (c) Hecht, S. M. In *The Many Faces of RNA*; Eggleston, D. S., Prescott, C. D., Pearson, N. D., Eds.; Academic Press: San Diego, 1998; pp 3–17.

⁽⁵⁾ Tanaka, W. J. Antibiot. 1977, 30, S-41.

^{(6) (}a) Umezawa, H.; Hori, S.; Sawa, T.; Yoshioka, T.; Takeuchi, T. J. Antibiot. **1974**, 27, 419. (b) Lazo, J. S.; Boland, C. J.; Schwartz, P. E. Cancer Res. **1982**, 42, 4026. (c) Sebti, S. M.; DeLeon, J. C.; Ma, L.-T.; Hecht, S. M.; Lazo, J. S. Biochem. Pharmacol. **1989**, 38, 141. (d) Schwartz, D. R.; Homanics, G. E.; Hoyt, D. G.; Klein, E.; Abernethy, J.; Lazo, J. S. Proc. Natl. Acad. Sci. U.S.A. **1999**, 96, 4680.

⁽⁷⁾ Tounekti, O.; Pron, G.; Belehradek, J. Jr.; Mir, L. M. Cancer Res. 1993, 53, 5462.

^{(8) (}a) Ashwell, G.; Harford, J. Annu. Rev. Biochem. **1982**, 51, 531. (b) Spiess, M. Biochemistry **1990**, 29, 10009.

glycoproteins and effects their internalization. Because of the rapid uptake mediated by the asialoglycoprotein receptor, this constitutes an attractive experimental system in which to study cell targeting.9 Recently, van Boom and co-workers¹⁰ reported the synthesis of galactosyl glycopeptides and generalized the structures required for the ligands to be recognized by the asialoglycoprotein receptor. It has also been reported¹¹ that both the hydrophobicity and the type of linker attached to the sugar moiety play important roles in ligand recognition.

Radionuclide conjugates of bleomycin have been shown to localize in tumors,¹² and individual naturally occurring bleomycin congeners have been reported to be distributed in a characteristic fashion to specific organs in mice.⁵ Despite the preparation of numerous analogues of bleomycin for defining the structural elements that are essential for DNA and RNA binding and cleavage,13 no analogous effort has addressed the molecular basis of organ or tumor targeting by bleomycin. Reported herein is the conjugation of BLM A₅ to a cluster galactoside capable of binding and internalization by the asialoglycoprotein receptor on liver cells.

The point of attachment of the cluster galactoside to bleomycin is clearly critical, in that the conjugate must be capable of binding and degrading those DNA and RNA targets responsible for the antitumor activity of the drug. The recent report¹⁴ that conjugation of BLM A₅ to a solid phase support had little effect on its ability to mediate DNA cleavage suggested that the C-terminal spermidine moiety might well constitute an appropriate site for conjugation of the cluster galactoside. Accordingly, we have prepared BLM conjugate 2 for study (Figure 1).

The synthesis of bleomycin conjugate 2 is outlined in Schemes 1 and 2. First, key intermediate 6 was synthesized (Scheme 1). Succinimidyl ester 3,15 prepared from monomethyl succinate in 86% yield, was treated with tris-(hydroxymethyl)aminomethane (4) in anhydrous DMF to afford trihydroxylated amide 5^{16} as a colorless solid in 50% yield. Thiomethylation of **5** with methyl sulfide and benzoyl peroxide¹⁷ in acetonitrile at 0 °C afforded tris methylthiomethyl derivative 6^{16} as a syrup in 38% yield.

Initial attempts to condense the tris(thiomethyl) derivative **6** with the known galactopyranosyl derivative **7**, ^{10a} prepared from commercially available galactose pentaacetate in four steps and 40% overall yield, in the presence of NIS/TfOH^{10a} failed to afford the desired trigalactosyl intermediate 8. Likewise, the use of MeOTf¹⁸ as a catalyst also failed to afford 8.

⁽¹²⁾ DeRiemer, L. H.; Meares, C. F.; Goodwin, D. A.; Diamanti, C. I. J. Med. Chem. 1979, 22, 1019.





Figure 1. Structures of bleomycin A_5 (1) and bleomycin A_5 glycoconjugate 2.

Compound 6 was then condensed with galactopyranosyl derivative 7 in the presence of N-bromosuccinimide¹⁹ to obtain the desired cluster galactoside $8^{20,21}$ (Scheme 2). Intermediate 8 was debenzoylated by treatment with sodium methoxide in methanol. Trigalactosyl derivative 9^{20} was obtained in 74% yield and a good state of purity following column chromatography on silica gel. Finally, methyl ester 9 was saponified using 1 M sodium hydroxide in 3:1 waterdioxane^{10b} to afford carboxylate **10**²⁰ in 48% yield.

The condensation of carboxylate 10 and the Cu(II) complex of bleomycin A_5^{22} was carried out via the agency of benzotriazol-1-yloxy tris(dimethylamino)phosphonium hexafluorophosphate (BOP reagent)²³ (Scheme 2). The crude



⁽⁹⁾ While the internalized ligands are initially contained within an endosomal compartment separate from the cytoplasm, and potentially susceptible to lysosome-mediated degradation, it is known that some of the ligands escape degradation and thus are potentially available for intracellular delivery.8

^{(10) (}a) Biessen, E. A. L.; Beuting, D. M.; Roelen, H. C. P. F.; van der Marel, G. A.; van Boom, J. H.; van Berkel, T. J. C. J. Med. Chem. 1995, 38, 1538. (b) Biessen, E. A. L.; Broxterman, H.; van Boom, J. H.; van Berkel, T. C. J. Med. Chem. 1995, 38, 1846.

^{(11) (}a) Lee, R. T.; Lin, P.; Lee, Y. C. Biochemistry 1984, 23, 4255. (b) Peter, M. G.; Boldt, P. C.; Niederstein, Y.; Peter-Katalinic, J. Liebigs Ann. Chem. 1990, 863. (c) Krebs, A.; Depew, W. T.; Szarek, W. A.; Hay, G. W.; Hronowski, L. J. J. Carbohydr. Res. 1994, 254, 257.



product was purified by C_{18} reversed-phase HPLC. The desired Cu(II) chelate of 2^{20} was recovered from the appropriate fractions by lyophilization as a bluish solid in 25% yield. Demetalation was effected by treatment of Cu(II)-2 with 15% Na₂EDTA. Purification by C_{18} reversed-phase HPLC afforded the desired BLM A₅ glycoconjugate 2 in 63% yield.

The ability of glycoconjugate **2** to degrade DNA was evaluated initially using supercoiled pSP64 plasmid DNA. As shown in Figure 2, significant cleavage was obtained



Figure 2. Relaxation of supercoiled pSP64 plasmid DNA by BLM A₅ glycoconjugate **2** in the presence of Fe²⁺. Twenty-five microliter reaction mixtures contained 300 ng of plasmid DNA in 10 mM Na cacodylate, pH 7.0, at 0 °C. After incubation for 15 min, the reaction mixtures were analyzed on a 1% agarose gel: lane 1, DNA alone; lane 2, 1 μ M Fe²⁺; lane 3, 1 μ M BLM **2**; lane 4, 0.5 μ M BLM **2** + 1 μ M Fe²⁺; lane 5, 1 μ M BLM **2** + 1 μ M Fe²⁺; lane 6, 2 μ M BLM **2** + 1 μ M Fe²⁺; lane 7, 1 μ M BLM **1**; lane 8, 0.5 μ M BLM **1** + 1 μ M Fe²⁺; lane 9, 1 μ M BLM **1** + 1 μ M Fe²⁺.

using $0.5 \,\mu\text{M}$ Fe(II)·BLM conjugate **2**, and greater than 50% conversion to form II (nicked circular) and III (linear duplex)

DNA resulted from the use of 1 μ M Fe(II)•BLM conjugate 2. Comparison with DNA relaxation mediated by Fe(II)• BLM A₅ (1) indicated that conjugate 2 was approximately one-half as potent as BLM A₅ in effecting DNA relaxation (cf. lanes 5 and 8, and lanes 6 and 9).

The sequence selectivity of DNA cleavage by Fe(II)·BLM glycoconjugate 2 is shown in Figure 3 in direct comparison

(15) Digenis, G. A.; Agha, B. J.; Tsuji, K.; Kato, M.; Shinogi, M. J. Med. Chem. 1986, 29, 1468.

(16) Compound **5**:¹H NMR (D₂O) δ 2.58–2.60 (s, 4H), 3.67 (s, 3H) and 3.73 (s, 6H); ¹³C NMR (D₂O) δ 28.7, 30.1, 51.8, 59.9, 61.5, 174.7 and 175.4; mass spectrum (chemical ionization, methane) 236 (M + H)⁺. Anal. Calcd for C₉H₁₇NO₆: C, 45.95; H, 7.28; N, 5.95. Found: C, 45.80; H, 7.12; N, 6.13. Compound 6: ¹H NMR (CDCl₃) δ 2.14 (s, 9H), 2.47 (t, 2H), 2.62 (t, 2H), 3.68 (s, 3H), 3.84 (s, 6H), 4.64 (s, 6H) and 5.92 (s, 1H); ¹³C NMR (CDCl₃) δ 14.1, 29.2, 31.5, 51.8, 58.9, 67.1, 75.9, 171.2 and 173.2; mass spectrum (chemical ionization, methane) 416 (M + H)⁺. Anal. Calcd for C₁₅H₂₉NO₆S₃: C, 43.36; H, 7.04; N, 3.37. Found: C, 43.62; H, 7.04; N, 3.33.

(17) Medina, J. C.; Salomon, M.; Kyler, K. S. Tetrahedron Lett. 1988, 29, 3773.

(18) Lonn, H. Carbohydr. Res. 1985, 139, 115.

(19) For a formally analogous transformation, see: Nicolaou, K. C.; Seitz, S. P.; Papahatjis, D. P. J. Am. Chem. Soc. **1983**, 105, 2430.

⁽¹³⁾ See, for example: (a) Owa, T.; Haupt, A.; Otsuka, M.; Kobayashi, S.; Tomioka, N.; Itai, A.; Ohno, M.; Shiraki, T.; Uesugi, M.; Sugiura, Y.; Maeda, K. *Tetrahedron* **1992**, *48*, 1193. (b) Hamamichi, N.; Natrajan, A.; Hecht, S. M. J. Am. Chem. Soc. **1992**, *114*, 6279. (c) Quada, J. C. Jr.; Levy, M. J.; Hecht, S. M. J. Am. Chem. Soc. **1993**, *115*, 12171. (d) Guajardo, R. J.; Hudson, S. E.; Brown, S. J.; Mascharak, P. K. J. Am. Chem. Soc. **1993**, *115*, 7971. (e) Kane, S. A.; Natrajan, A.; Hecht, S. M. J. Biol. Chem. **1994**, *269*, 10899. (f) Boger, D. L.; Honda, T.; Menezes, R. F.; Colletti, S. L.; Dang, Q.; Yang, W. J. Am. Chem. Soc. **1994**, *116*, 82. (g) Boger, D. L.; Honda, T.; Menezes, R. F.; Colletti, S. L. J. Am. Chem. Soc. **1995**, *117*, 7344. (i) Katano, K.; An, H.; Aoyagi, Y.; Overhand, M.; Sucheck, S. J.; Stevens, W. C. Jr.; Hess, C. D.; Zhou, X.; Hecht, S. M. J. Am. Chem. Soc. **1998**, *120*, 11285.

⁽¹⁴⁾ Abraham, A. T.; Zhou, X.; Hecht, S. M. J. Am. Chem. Soc. 1999, 121, 1982.



Figure 3. Strand scission of a linear DNA duplex by BLM A₅ conjugate **2** in the presence of Fe²⁺. A 5'-³²P end labeled 158-bp DNA duplex was incubated with Fe(II)·BLM at 4 °C for 15 min and then analyzed on a 10% denaturing polyacrylamide gel: lane 1, 0.5 μ M BLM **2**; lane 2, 0.5 μ M BLM **2** + 20 μ M Fe²⁺; lane 3, 0.25 μ M BLM **1**; lane 4, 0.25 μ M BLM **1** + 20 μ M Fe²⁺; lane 5, DNA alone; lane 6, 20 μ M Fe²⁺; lane 7, 1 μ M BLM **2**, lane 8, 1 μ M BLM **2** + 20 μ M Fe²⁺.

with that mediated by Fe(II)•BLM A₅ (1). The substrate was a 158-base pair 5'-³²P end labeled DNA duplex that has been

employed previously as a substrate for Fe(II)•BLM.¹⁴ As is clear from the figure, Fe(II)•BLM glycoconjugate **2** exhibited the same sequence selectivity of DNA cleavage as BLM A₅ (**1**); the potency of cleavage by **2** was only 2–3-fold less than that obtained with **1**. Thus, the introduction of a glycosylated substituent at the C-terminus of BLM A₅ had no effect on the sequence selectivity of DNA cleavage and only a modest effect on cleavage potency, consistent with observations made for BLM derivatives conjugated to solid supports via the C-terminus.¹⁴

Bleomycin derivative 2 was tested for its cytotoxic potential in direct comparison with BLM A_5 (1). In preliminary experiments, 2 exhibited cytotoxicity comparable to BLM A_5 toward cultured human epidermal carcinoma A253 cells. Experiments designed to evaluate the ability of BLM glycoconjugate 2 to interact selectively with liver cells are underway and will be reported in due course.

Acknowledgment. We thank Dr. John Lazo and Katharine Pestell, University of Pittsburgh, for the cytotoxicity data. This work was supported by NIH Research Grants CA76297 and CA77284, awarded by the National Cancer Institute.

OL010015H

(22) Prepared by admixture of CuSO₄·5H₂O to an aqueous solution of bleomycin A₅, thus forming a derivative in which only the C-terminal substituent could undergo acylation.¹⁴

(23) (a) Castro, B.; Dormoy, J. R.; Evin, G.; Selve, C. *Tetrahedron Lett.* **1975**, *16*, 1219. (b) Castro, B.; Dormoy, J. R.; Dourtoglou, B.; Evin, G.; Selve, C.; Ziegler, J. C. *Synthesis* **1976**, 751. (c) Castro, B.; Evin, G.; Selve, C.; Seyer, R. *Synthesis* **1977**, 413.

⁽²⁰⁾ Compound 8: ESI-MS, m/z 2610.3 (M + Na)⁺ (C₁₃₈H₁₄₉O₄₈NNa requires 2610.9). Compound 9:¹³C NMR (D₂O) δ 28.6, 30.2, 51.8, 60.5, 65.8, 66.3, 68.2, 68.9–69.3 [C from tris(tetraethylene glycol)], 70.3, 72.2, 74.7, 95.0, 102.4, 173.9 and 175.2; ESI-MS, m/z 1362.8 (M + Na)⁺ (C₅₄H₁₀₁O₃₆NNa requires 1362.6). Compound **10**: ESI-MS, m/z 1348.7 (M + Na)⁺ (C₅₃H₉₉O₃₆NNa requires 1348.6). Compound Cu(II)-**2**: ESI-MS, m/z 2809.4 (M)⁺ (C₁₁₀H₁₈₆O₅₆N₂₀S₂Cu requires 2810.1). Compound **2**: ESI-MS, m/z 2770.3 (M + Na)⁺ (C₁₁₀H₁₈₆O₅₆N₂₀S₂Na requires 2770.2).

⁽²¹⁾ Compound 8 was obtained as a mixture, contaminated predominantly with compounds resulting from the condensation of only one or two molecules of galactoside 7 with intermediate 6. A sample greatly enriched in 8 was obtained in 50% yield following chromatography on silica gel and then on Sephadex LH-20.